

Recent Advances in the Green Synthesis of Gold and Silver Nanostructures for Augmented Anti-Microbial Activity

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Received: June 2021

Revised: September 2021

Accepted: December 2021

DOI: 10.22068/ijmse.2252

Abstract: Green synthesis refers to the synthesis of nanoparticles using plants and microorganisms. It is preferred over conventional methods because it is sustainable, eco-friendly, cost-effective, and fast. The phytochemicals and enzymes present in plants and microorganisms respectively acts as the reducing and capping agent for the synthesis of nanoparticles. Phytochemicals and enzymes have the ability to reduce precursor metal ions into nanoparticles. As the conventional methods involve the use of high energy and toxic chemicals which are harmful to both environment and organisms, these synthesis methods are discouraged. Of the nanoparticles, gold nanoparticles (AuNPs) and silver nanoparticles (AgNPs) have gained lots of attention owing to their multiple applications and less toxicity. In addition, various in-vitro studies have reported the antimicrobial activity of AgNPs and AuNPs against various microbes. This particular review portrays the methods of nanoparticles synthesis, components of green synthesis, mechanism of green synthesis, antimicrobial activity, other applications and various factors affecting the green synthesis of AgNPs and AuNPs.

Keywords: Green synthesis, Nanoparticles, AuNPs, AgNPs, Plants, Microorganisms, Antimicrobial activity.

1. INTRODUCTION

The field of nanotechnology has gained tremendous importance in modern materials science [1]. The term “nanotechnology” was first coined by the American physicist Richard Feynman in 1959 [2]. However, the history dates back to the 9th century, when the artisans of Mesopotamia age utilized the gold nanoparticles (less than 100 nm) to produce glittering effects on pots and other utensils [3]. In general, nanotechnology refers to the application of science to control matter at the molecular stage [4]. It is also gaining importance in other areas of modern science like optics, mechanics, chemical industry, electronics, aeronautics, biomedical sciences, drug-gene delivery, catalysis, optoelectronic devices, energy science, and photo-electrochemical applications [4].

Recently, the utilization of nanoparticles, especially gold and silver nanoparticles, has accumulated significant spotlight in nanoscience owing to its multiple physical, chemical, and biological applications [5]. And nanoparticles are synthesized by two major processes top-down and bottom-up processes. The conventional synthesis methods include physical and chemical methods, which are highly discouraged owing to its high production cost, extensive usage of energy,

production of chemical waste products that contaminate the environment, and excessive production of hazardous and toxic chemicals that are harmful to the organisms.

Due to its major setbacks, the green synthesis of nanoparticles has gained attention over conventional methods. The green synthesis method is safe, reliable, efficient, eco-friendly, and sustainable [6] (Fig. 1). Multiple applications of nanoparticles are credited for their very small size (<100 nm), distinct morphology, and extensive surface area [7-9].

Green synthesis is defined as reducing precursor metal or metal oxide ions into their respective nanoparticles by using the biological components. It is further divided into two major categories, plant-based synthesis and microorganism-based synthesis [6]. However, the exact mechanism of how nanoparticles are synthesized using the plant is not known. Still, it is claimed that phytochemicals present in the plant extract like ketones, aldehydes, flavones, amides, terpenes, carboxylic acids, phenol, and ascorbic acid reduce precursor metal salts into metal nanoparticles [10, 11]. The phytochemical present in plant extracts acts as both reducing and capping agents, external reducing and capping agents are not needed. While the intracellular enzyme present in microorganisms reduces the

metal ions into their individual metal nanoparticles [12, 13, 14]. Green synthesis is dependent on various factors like the technique of preparation, the concentration of precursor metal salt, pH, temperature, pressure, time, preparation cost, shape, size, pore size, environment, the proximity of other nanoparticles, and other factors [15]. The NPs produced by green synthesis are used in nanomedicine, cancer treatment, antimicrobial agent, DNA analysis, targeted drug delivery, gene treatment, biosensors, and medical imaging [6, 16].

NPs are also associated with certain limitations or challenges such as stability in the hostile environment, bioaccumulations, toxicity, need for skilled operators, and the lack of understanding of the fundamental mechanism and modeling factors [6].

This review focuses on different methods of nanoparticles synthesis, types of green synthesis, its components, mechanism of plant and microorganism based-synthesis, factors affecting green synthesis, anti-microbial activity, other biomedical applications, and recent advances in the green synthesis of Au and Ag nanoparticles.

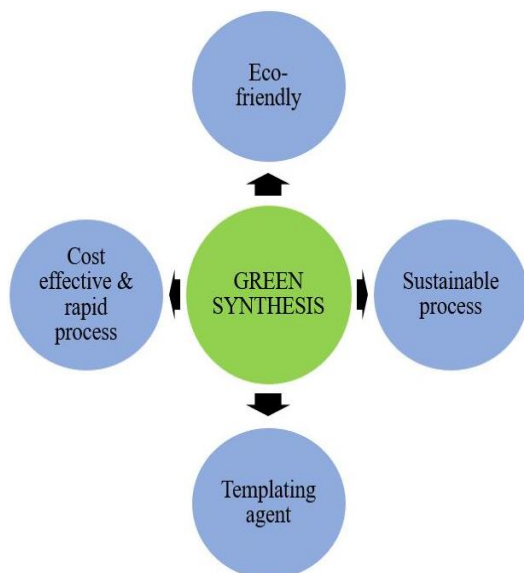


Fig. 1. Advantages of green synthesis

1.1. Different methods of NPs synthesis

The synthesis of nanoparticles is done by two major approaches, the top-down approach and the bottom-up approach [6]. The top-down approach involves the breaking down the bulk material into nanosized structures or particles. In a top-down

approach, chemical and physical methods like chemical etching, laser ablation, mechanical milling, sputtering, electro-explosion, thermolysis, lithography, and pyrolysis reduces the precursor metal or metal oxide salt ions into nanoparticles [17]. The methodologies used in the top-down approach are related to major drawback as they resulted in the formation of irregular surfaced nanoparticles. Which in turn results in modified physical and chemical properties. This approach also requires high energy, i.e., temperature and pressure and high production costs [18]. Different methods of nanoparticles synthesis are shown in Fig. 2.

The bottom-up approach is also known as the self-assembly approach. In this, the synthesis of nanoparticles is done by assembling smaller molecules or atoms, thus forming a larger structure (<100 nm).

The methods used in this approach include chemical vapor deposition, laser pyrolysis, spray pyrolysis, sol-gel process, atomic molecular condensation, and aerosol process [19]. It has gained more importance over the top-down approach as it can produce the NPs with the homogenous chemical composition and less surface irregularity at a cheaper rate. However, this approach is also associated with its drawbacks. It involves reducing agents, capping agents, non-polar organic solvent and hazardous chemicals that produce chemical waste and toxic chemicals contaminating reducing agents, capping agents, non-polar organic solvent, and hazardous chemicals that produce chemical waste toxic chemicals contaminating the environment and posing biological hazards. Because of the drawbacks mentioned above, its use in medical and biomedical applications is restricted [20].

Due to the major drawbacks of the two approaches, the green synthesis of NPs has caught the attention of researchers as it is safe, eco-friendly, sustainable, efficient, and cheap compared to conventional chemical and physical methods [6].

Green synthesis involves using biological components to reduce the precursor metal salt ions into nanoparticles. The phytochemicals in plant extract and intracellular enzymes in an organism act as reducing and capping agents [12, 13, 14].

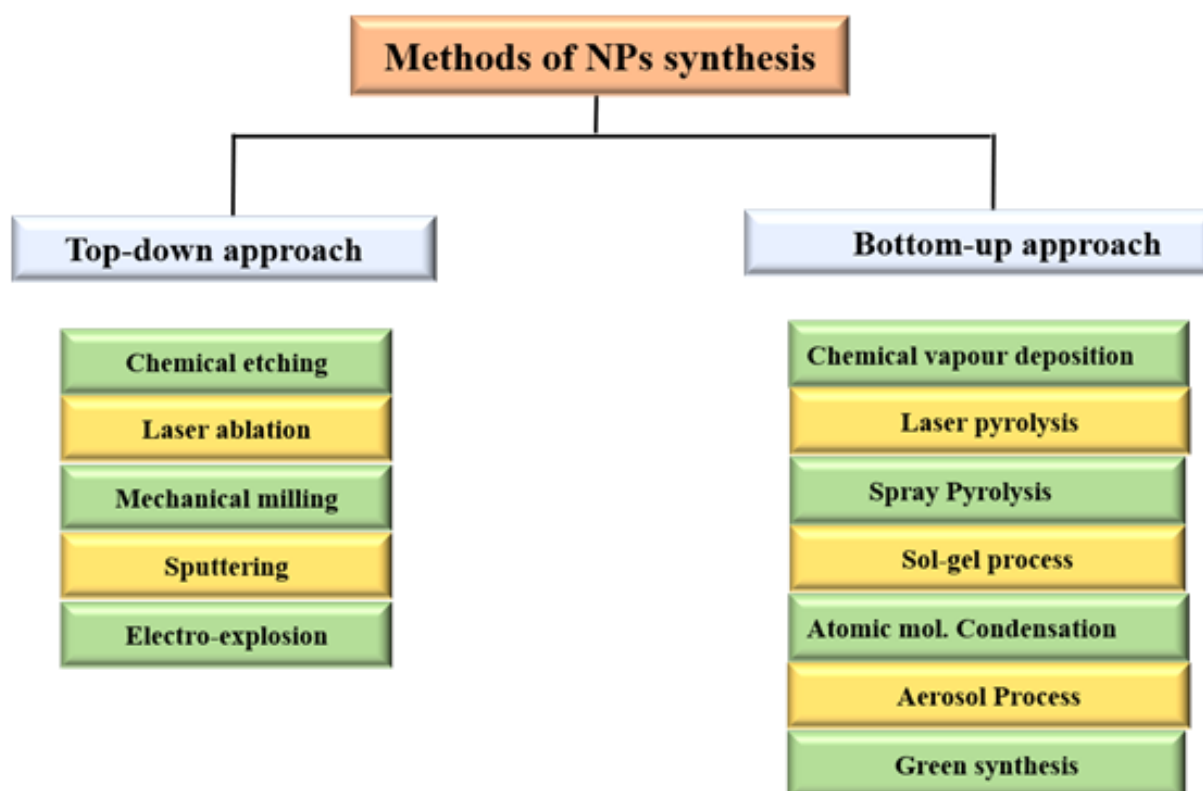


Fig. 2. Different methods of nanoparticles synthesis

1.2. Components used for green synthesis of NPs

The conventional method of green synthesis involves using a highly toxic reducing agent stabilizing agent, and requires a high amount of energy for the synthesis. In addition, it also produces lots of chemical waste, and toxic chemicals that are both detrimental to humans and the environment [18]. While green synthesis of nanoparticles is eco-friendly, rapid, and requires less energy, capable of large-scale productions, cheap and sustainable. And the components used for green synthesis are broadly divided into two main categories plant and microorganism [6].

Different components of green synthesis are shown in Fig. 3.

1.2.1. Plant

Plants have the innate potential to reduce precursor metal salt ions into respective nanoparticles. Various parts of the plant such as leaf, flower, fruit, bark, stem, and root extract are used to reduce metal and metal oxide ions.

Since this method is simple, eco-friendly, efficient, sustainable, and cost-effective, it is preferred over chemical and physical methods [6].

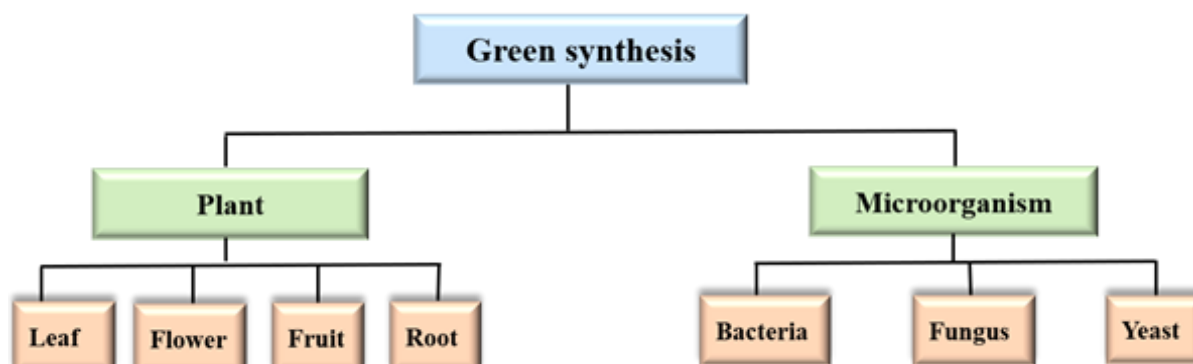


Fig. 3. Different biological components of green synthesis of nanoparticles.

Plant extractions contain biological molecules such as enzymes, flavones, ascorbic acid, oxalic acid, carboxylic acid, terpenoids, glycosides, carbohydrates, polyols, thiamine, proteins, amino acids, polysaccharides, flavonoids and vitamins, which are environmentally benign. These molecules play a major role in reducing and stabilizing metal nanoparticles [10, 11].

Recently, researchers have shifted their focus to synthesise nanoparticles using plant as the phytochemical present in it acts as both reducing and capping agent along with biomolecules present in it whereby use of toxic chemicals are not required [12, 13, 14]. Many researches have been done to date on ex-vivo (outside living plant) synthesis of NPs, while synthesis can also be done in-vivo (inside living plant) by the reduction of absorbing precursor metal or metal oxide ions into soluble salt.

Gold and silver nanoparticles have gained more attention due to their efficiency and diverse applications in various fields. And the synthesis of Au and Ag nanoparticles was first investigated in plant extract-based synthesis [6]. Different plants used for the synthesis of Au and Ag nanoparticles and morphology and size are shown in table 1 and table 2, respectively.

1.2.2. Microorganism

This method involves using biological components, namely bacteria, fungus, and yeast. Other studies reported the synthesis via viruses, agricultural waste, food waste, microbes, and biopolymers [85].

1.2.2.1 Bacteria

The use of bacteria has gained importance in various fields like genetic engineering, targeted-drug delivery, bioremediation, and bioleaching [86]. Bacteria can reduce the precursor metal salt ions into metal nanoparticles. Bacteria as the biological components for synthesis has been a promising biological reductant for the preparation of nanoparticles [87].

Different varieties of bacterial species are made into use for the synthesis. Of various species used, actinomycetes and prokaryotic bacteria have been prominent for nanoparticle synthesis due to its versatility, easy manipulation, and efficiency [88].

Commonly used bacterial species for the synthesis of AgNPs includes *Lactobacillus casei*,

E. coli, *Aeromonas* sp. SH10, *Phaeocystis antarctica*, *Arthrobacter gangotriensis*, *Geobacter* spp., *Corynebacterium* sp. SH09 and *Shewanella oneidensis*. Table 3 shows the bacterial species used to synthesize AgNPs with distinct sizes and morphology. For the preparation of AuNPs, different bacterial species such as *Desulfovibrio desulfuricans*, *E. coli* DH5a, *Rhodopseudomonas capsulate*, *Bacillus megaterium* D01, *Bacillus subtilis* 168, *Plectonema boryanum* UTEX 485, and *Shewanella alga* have been widely used. Table 4 shows different bacterial species used for the synthesis of AuNPs with distinct sizes and morphology.

1.2.2.1 Fungi

Besides bacteria, fungi also served as the efficient component for the synthesis of gold and silver nanoparticles with definite morphologies and sizes. Due to a variety of intracellular enzymes, fungi serve as the better synthesis component of metal and metal oxide nanoparticles [89]. The study also reported that competent fungi can be better than bacteria in terms of the amount and efficiency of synthesis of NPs [90].

Fungi-mediated synthesis of NPs has many advantages over other microorganisms due to the presence of enzymes, reductants, and proteins on the surface of their cell [91]. However, the actual mechanism of fungi-mediated synthesis of NPs is not known, but researchers and scholars believe that formation is due to enzymatic reduction in the cell wall or inside the fungal cell. Tables 3 and 4 summarize published research on the synthesis of silver and gold nanoparticles using fungi.

1.2.2.2 Yeast

Yeasts are single-celled microorganisms present in eukaryotic cells. A total of 1500 yeast species have been identified [92]. The successful synthesis of nanoparticles/ nanomaterials via yeast has been reported by numerous research groups. The biosynthesis of silver and gold nanoparticles by a silver-tolerant yeast strain and *Saccharomyces cerevisiae* broth has been reported [93]. Table 3 and 4 show the varieties of yeast species used to synthesize AgNPs and AuNPs with distinct size and morphology.

Table 1. Synthesis of AuNPs using different plant sources

| Sl. No | Plant source | Part Used | Morphology | Size (nm) | Ref. |
|--------|-------------------------------------------------|-------------|---------------------------------------------------------------------|--------------|--------------------------------------|
| 1 | <i>Abelmoschus esculentus</i> | Seed | Spherical | 45– 75 | (Jayaseelan <i>et al.</i> , 2013) |
| 2 | <i>Aerva lanata</i> | Leaf | Mostly Spherical | 17.97 | (Joseph <i>et al.</i> , 2015) |
| 3 | <i>Chenopodium album</i> | Leaf | Spherical | Various | (Dwivedi <i>et al.</i> , 2010) |
| 4 | <i>Cinnamomum zeylanicum</i> | Bark, Leaf | Nanoprisms & Spheres | 25 | (Smitha <i>et al.</i> , 2009) |
| 5 | <i>Eucommia ulmoides</i> | Bark | Spherical | 16.4 | (Guo <i>et al.</i> , 2015) |
| 6 | <i>Curcuma pseudomontana</i> | Seed | Spherical | 20 | (Muniyappan & Nagarajan, 2014) |
| 7 | <i>Geranium</i> | Leaf | Variable | 8– 20 | (Franco-Romano <i>et al.</i> , 2014) |
| 8 | <i>Hibiscus rosa sinensis</i> | Leaf | Variable | Various | (Philip, 2010) |
| 9 | Grape waste | Fruit | Spherical | 20– 25 | (Krishnaswamy <i>et al.</i> , 2014) |
| 10 | <i>Hovenia dulcis</i> | Fruit | Variable | 15– 20 | (Basavegowda <i>et al.</i> , 2014) |
| 11 | <i>J. sambac leaves</i> | Leaves | Spherical | 20– 50 | (Yallappa <i>et al.</i> , 2015) |
| 12 | <i>Krishna tulsi</i> | Leaf | Spherical | 30 | (Philip & Unni, 2011) |
| 13 | <i>Lansium domesticum</i> | Fruit peel | Triangular, Hexagonal | 20– 40 | (Shankar <i>et al.</i> , 2014) |
| 14 | <i>Mangifera indica</i> | Leaf | Spherical | 17– 20 | (Philip, 2010) |
| 15 | <i>Magnolia kobus</i> | Leaf | Variable | 5– 300 | (Song <i>et al.</i> , 2009) |
| 16 | Maple leaf pine needle | Leaf | Spherical | < 100 | (Krishnaswamy & Orsat, 2015) |
| 17 | <i>Morinda citrifolia</i> | Root | Spherical | 12.17– 38.26 | (Suman <i>et al.</i> , 2014) |
| 18 | <i>Mentha piperita</i> | - | Spherical | 150 | (MubarakAli <i>et al.</i> , 2011) |
| 19 | <i>Cymbopogon sp</i> | Leaf | Spherical, triangular | 200– 500 | (Shankar <i>et al.</i> , 2005) |
| 20 | <i>Dioscorea bulbifer</i> | Tuber | Spherical | 11– 30 | (Ghosh <i>et al.</i> , 2011) |
| 21 | <i>Emblica officinalis</i> | Fruit | Polyhedral | 15– 25 | (Ankamwar <i>et al.</i> , 2005) |
| 22 | <i>Tamarindus indica</i> | Leaf | Triangular | 20– 40 | (Ankamwar <i>et al.</i> , 2005) |
| 23 | <i>Cinnamomum camphora</i> | Leaf | Spherical, triangular | 55– 80 | (Huang <i>et al.</i> , 2007) |
| 24 | <i>Sesbania</i> | - | Spherical | 6– 20 | (Sharma <i>et al.</i> , 2007) |
| 25 | <i>Cymbopogon flexuosus</i> | - | Triangular | - | (Shankar <i>et al.</i> , 2004) |
| 26 | <i>Salvia officinalis</i> | Whole plant | Spherical, triangular, truncated triangles, pentagons and hexagons. | 4– 72 | (Elia <i>et al.</i> , 2014) |
| 27 | <i>Pelargonium graveolens</i> | Leaf | Spherical rods, flat, sheets and triangular | 21– 70 | (Shankar <i>et al.</i> , 2003) |
| 28 | <i>Lippia citriodora</i> | Leaf | Spherical, triangular, truncated triangles, pentagons and hexagons | 2.6– 50 | (Shankar <i>et al.</i> , 2003) |
| 30 | <i>Phoenix dactylifera</i> | Leaf | Spherical | 35– 45 | (Zayed <i>et al.</i> , 2014) |
| 31 | <i>Pistacia integerrima</i> | gall | Spherical | 20– 200 | (Islam <i>et al.</i> , 2019) |
| 32 | <i>Pogostemon benghalensis</i> (B) O. Ktz. leaf | Leaf | Variable | 10– 50 | (Paul <i>et al.</i> , 2015) |
| 33 | <i>Solanum nigrum</i> | Leaf | Spherical | 50 | (Muthuvel <i>et al.</i> , 2014) |
| 34 | <i>Terminalia arjuna</i> | Leaf | Spherical | 20– 50 | (Gopinath <i>et al.</i> , 2013) |
| 35 | <i>Zingiber officinale</i> | Rhizome | Spherical | 5– 15 | (Kumar <i>et al.</i> , 2011) |

Table 2. Synthesis of AgNPs using different plant sources

| Sl. No. | Plant source | Part Used | Morphology | Size (nm) | Ref |
|---------|-------------------------------|---------------|----------------------|-----------|---------------------------------------|
| 1 | <i>Aerva lanata</i> | - | Mostly Spherical | 18.62 | (Joseph <i>et al.</i> , 2015) |
| 2 | <i>Artemisia nilagirica</i> | Whole plant | Predominantly Square | 70–90 | (Vijayakumar <i>et al.</i> , 2013) |
| 3 | Banana peel | Fruit | Spherical | 23.7 | (Ibrahim, 2015) |
| 4 | Beetroot | Root | Spherical | 15 | (Bindu <i>et al.</i> , 2015) |
| 5 | <i>Boerhavia diffusa</i> | Whole plant | Spherical | 25 | (Kumar <i>et al.</i> , 2014) |
| 6 | <i>Cinnamon zeylanicum</i> | Bark | Variable | 31–40 | (Sathishkumar <i>et al.</i> , 2009) |
| 7 | <i>Cocos nucifera</i> | Inflorescence | Spherical | 22 | (Mariselvam <i>et al.</i> , 2014) |
| 8 | <i>Coriandrum sativum</i> | seed | Spherical | 13.09 | (Nazeruddin <i>et al.</i> , 2014) |
| 9 | <i>Dalbergia spinosa</i> | Leaf | Spherical | 18 | (Muniyappan <i>et al.</i> , 2014) |
| 10 | <i>Delonix elata</i> | Leaf | Spherical | 35–45 | (Sathiya & Akilandeswari, 2014) |
| 11 | <i>Embllica officinalis</i> | Fruit | Spherical | 15 | (Ramesh <i>et al.</i> , 2015) |
| 12 | <i>Eucalyptus</i> | Leaf | Spherical | 40–60 | (Pourmortazavi <i>et al.</i> , 2015) |
| 13 | <i>Ficus carica</i> | Leaf | Spherical | 13 | (Ulug <i>et al.</i> , 2015) |
| 14 | <i>Gloriosa superba</i> | Leaf | Spherical | 10–25 | (Ashokkumar <i>et al.</i> , 2013) |
| 15 | <i>Hibiscus rosa sinensis</i> | - | Variable | various | (Philip, 2010) |
| 16 | <i>A. indicum</i> | Leaf | Spherical | 7–17 | (Ashokkumar <i>et al.</i> , 2015) |
| 17 | <i>J. sambac</i> | Leaf | Spherical | 20–50 | (Yallappa <i>et al.</i> , 2015) |
| 18 | Krishna tulsi | Leaf | Spherical | 10–20 | (Philip & Unni, 2011) |
| 19 | <i>Lansium domesticum</i> | Fruit peel | Spherical | 10–30 | (Shankar <i>et al.</i> , 2014) |
| 20 | <i>Lantana camara</i> leaf | Leaf | Spherical | 20–34 | (Ajitha <i>et al.</i> , 2015) |
| 21 | <i>Lippia citriodora</i> | - | Spherical | 15–30 | (Cruz <i>et al.</i> , 2010) |
| 22 | <i>Mangifera indica</i> | Leaf | Spherical | 20 | (Philip, 2011) |
| 23 | Maple leaf | Leaf | Spherical | 1–100 | (Vivekanandhan <i>et al.</i> , 2014) |
| 24 | Neem gum | Gum | Spherical | < 30 | (Velusamy <i>et al.</i> , 2015) |
| 25 | <i>Papaver somniferum</i> | - | Spherical | 3200–7600 | (Vijayaraghavan <i>et al.</i> , 2012) |
| 26 | <i>Piper longum</i> | Leaf | Spherical | 17.6–41 | (Jacob <i>et al.</i> , 2012) |
| 27 | <i>Piper nigrum</i> | - | Spherical | 32–1 | (Mohapatra <i>et al.</i> , 2015) |
| 28 | Tea leaf | Leaf | Spherical | 20–90 | (Sun <i>et al.</i> , 2014) |
| 29 | <i>Tephrosia purpurea</i> | Leaf | Spherical | 20 | (Ajitha <i>et al.</i> , 2014) |
| 30 | <i>Tribulus terrestris</i> | - | Spherical | 16–28 | (Gopinath <i>et al.</i> , 2012) |
| 31 | <i>Ziziphus jujuba</i> | Leaf | Variable | 20–30 | (Gavade <i>et al.</i> , 2015) |

Table 3. Synthesis of AgNPs using different microbial culture (94)

| Sl.no | Microbial culture | Morphology | Size (nm) |
|-----------------|-------------------------------------|------------|-----------|
| Bacteria | | | |
| 1 | <i>Bacillus licheniformis</i> | Irregular | 50 |
| 2 | <i>Bacillus cereus</i> | Spherical | 4–5 |
| 3 | <i>Brevibacterium casei</i> | Spherical | 10–50 |
| 4 | <i>Corynebacterium glutamicum</i> | Irregular | 5–50 |
| 5 | <i>Escherichia coli</i> | Irregular | 50 |
| 6 | <i>Pediococcus pentosaceus</i> | Irregular | < 100 |
| Fungi | | | |
| 7 | <i>Coriolus versicolor</i> | Spherical | 25–75 |
| 8 | <i>Cladosporium cladosporioides</i> | Spherical | 10–100 |
| 9 | <i>Aspergillus flavus</i> | Spherical | 8.92 |
| 10 | <i>Aspergillus terreus</i> | Spherical | 1–20 |
| 11 | <i>Fusarium oxysporum</i> | Spherical | 5–50 |
| 12 | <i>Fusarium oxysporum</i> | Irregular | 5–15 |
| 13 | <i>Fusarium oxysporum</i> | Spherical | 20–50 |
| 14 | <i>Fusarium semitectum</i> | Spherical | 10–60 |
| 15 | <i>Macrophomina phaseolina</i> | Spherical | 5–40 |
| 16 | <i>Penicillium fellutanum</i> | Spherical | 5–25 |
| 17 | <i>Penicillium nalgioense</i> | Spherical | 25 ± 2.8 |
| 18 | <i>Aspergillus fumigatus</i> | Spherical | 5–25 |
| 19 | <i>Phaenerochaete chrysosporium</i> | Pyramidal | 50–200 |
| 20 | <i>Phoma glomerata</i> | Spherical | 60–80 |

| | | | |
|--------------|-------------------------------|----------------------|-------|
| 21 | <i>Pleurotus sajor-caju</i> | Spherical | 30.5 |
| 22 | <i>Trichoderma reesei</i> | Spherical | 5–50 |
| 23 | <i>Trichoderma viride</i> | Spherical | 5–40 |
| 24 | <i>Trichoderma viride</i> | Spherical | 5–40 |
| 25 | <i>Trichoderma viride</i> | Irregular | 2–4 |
| 26 | <i>Trichoderma asperellum</i> | Spherical | 13–18 |
| 27 | <i>Verticillium sp.</i> | Spherical | 5–50 |
| Yeast | | | |
| 28 | Yeast | Irregular, Polygonal | 9–25 |

Table 4. Synthesis of AuNPs using different microbial culture [94]

| Sl. No | Microbial culture | Morphology | Size (nm) |
|-----------------|-----------------------------------------------|--------------------------|------------|
| Bacteria | | | |
| 1 | <i>Brevibacterium casei</i> | Spherical | 10–50 |
| 2 | <i>Escherichia coli</i> | Triangular, Hexagons | 20–30 |
| 3 | <i>Pediococcus pentosaceus</i> | Spherical | < 100 |
| 4 | <i>Plectonema boryanum</i> | Octahedral | 10–6000 |
| 5 | <i>Plectonema boryanum</i> | Cubic | 10–25 |
| 6 | <i>Pseudomonas aeruginosa</i> | Irregular | 15–30 |
| 7 | <i>Rhodococcus sp.</i> | Spherical | 5–15 |
| 8 | <i>Rhodopseudomonas capsulate</i> | Spherical | 10–20 |
| 9 | <i>Shewanella algae</i> | Irregular | 10–20 |
| 10 | <i>Shewanella oneidensis</i> | Spherical | < 100 |
| 11 | <i>Ureibacillus thermosphaericus</i> | Irregular | 50–70 |
| Fungi | | | |
| 12 | <i>Aureobasidium pullulans</i> | Spherical | 29 ± 6 |
| 13 | <i>Aspergillus clavatus</i> | Triangular, Spherical | 24.4 ± 11 |
| 14 | <i>Alternaria alternata</i> | Spherical, Triangular | 12 ± 5 |
| 15 | <i>Aspergillus sydowii</i> | Spherical | 8.7–15.6 |
| 16 | <i>Aspergillus niger</i> | Polydispersed | 10–20 |
| 17 | <i>Aspergillus niger</i> | Spherical | 12.8 ± 5.6 |
| 18 | <i>Fusarium semitectum</i> | Spherical | 10–80 |
| 19 | <i>Fusarium oxysporum</i> | Spherical | 2–50 |
| 20 | <i>Hansenula anomala</i> | Irregular | 14 |
| 21 | <i>Helminthosporium solani</i> | Variable | 2–70 |
| 22 | <i>Hormoconis resinae</i> | Spherical | 3–20 |
| 23 | <i>Neurospora crassa</i> | Spherical | 32 |
| 24 | <i>Penicillium rugulosum</i> | Spherical | 20–40 |
| 25 | <i>Aspergillus oryzae</i> var. <i>viridis</i> | Mostly Spherical | 10–60 |
| 26 | <i>Penicillium brevicompactum</i> | Various | 10–60 |
| 27 | <i>Penicillium sp</i> | Spherical | 30–50 |
| 28 | <i>Phanerochaete chrysosporium</i> | Spherical | 10–100 |
| 29 | <i>Cylindrocladium floridanum</i> | Spherical | 5–35 |
| 30 | <i>Cylindrocladium floridanum</i> | Spherical | 19.05 |
| 31 | <i>Coriolus versicolor</i> | Spherical | 20–100 |
| 32 | <i>Rhizopus oryzae</i> | Spherical | 16–25 |
| 33 | <i>Colletotrichum sp</i> | Spherical | 8–40 |
| 34 | <i>Candida albicans</i> | Spherical | 20–40 |
| 35 | <i>Sclerotium rolfsii</i> | Spherical | 25.2 ± 6.8 |
| 36 | <i>Candida albicans</i> | Monodispersed, Spherical | 5 |
| 37 | <i>Epicoccum nigrum</i> | - | 5–50 |

| | | | |
|--------------|---------------------------------|-----------------------|--------|
| 38 | <i>Trichoderma koningii</i> | Spherical | 30–40 |
| 39 | <i>Trichoderma koningii</i> | Spherical | 10–14 |
| 40 | <i>Verticillium luteoalbum</i> | Irregular | < 100 |
| 41 | <i>Volvariella volvacea</i> | Triangular, Spherical | 20–150 |
| 42 | <i>Verticillium sp</i> | Irregular | < 100 |
| 43 | <i>Verticillium sp</i> | Spherical | 20 ± 8 |
| 44 | <i>Yarrowia lipolytica</i> | Various | < 100 |
| 45 | <i>Yarrowia lipolytica</i> | Triangles | 15 |
| Yeast | | | |
| 46 | <i>Yeast</i> | Irregular, Polygonal | 9–25 |
| 47 | <i>Candida utilis</i> | Irregular | < 100 |
| 48 | <i>Saccharomyces cerevisiae</i> | Spherical | 15–20 |

1.3. Probable mechanism of synthesis of Ag and Au Nanoparticles

1.3.1. Microorganism based

Various mechanisms for the synthesis of NPs using microorganisms have been reported. However, when simplified, the mechanism revolves around a similar concept where precursor metal or metal oxide salt ions are captured on the surface or inside microbial cells. With the assistance of extra or intracellular enzymes, the captured ions are then reduced into respective nanoparticles. One of the studies reported the synthesis of Au and Ag NPs via *Verticillium sp* [95], and the hypothesis stated that initially, Au and Ag ions were captured on the fungal cell surface with the help of electrostatic interaction between ions and negatively charged cell wall enzyme. Then, the captured gold and silver ions were bio-reduced into gold and silver nuclei, which then subsequently grew into nanoparticles. The two main aspects in the biosynthesis of nanoparticles are nicotinamide adenine dinucleotide (NADH) and NADH-dependent nitrate reductase. Another study demonstrated the synthesis of AgNPs via *B. licheniformis*, which concluded that nitrate reductase was responsible for the synthesis [96]. However, the exact mechanism for the production of metal salt ions and metal and metal oxide nanoparticles is still not known. The general mechanism of synthesis is shown in fig. 4.

1.3.2. Plant based synthesis

After maintaining the conditions suitable for the synthesis, such as phytochemicals, metal salt concentration, phytochemical concentration, pH, and temperature, the extract of the plant is mixed

with metal precursor salt ions solution whereby synthesizing nanoparticles [96]. By maintaining the aforementioned conditions, the rate of nanoparticle formation, yield, and stability of nanoparticles can be controlled [97]. In comparison to fungi and bacteria, which need a longer incubation period, phytochemicals found in plant leaf extracts have an uncanny ability to reduce metal ions in a much shorter time [98].

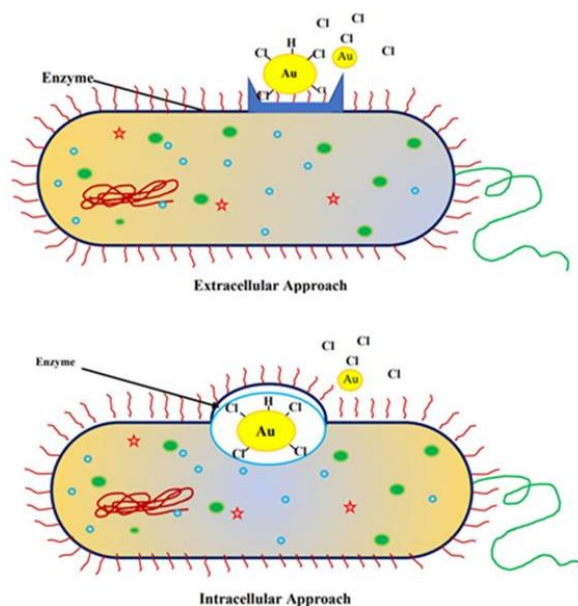


Fig. 4. Synthesis of AuNPs via different approaches using bacteria

In addition, plant leaf extracts are considered to be a good and safe source for metal and metal oxide nanoparticle synthesis. The plant extract has the potential to act as both a reducing and stabilizing agent for the synthesis process [12, 13, 14]. Within a few minutes, the reaction is completed. When the precursor solutions are

mixed, the biochemical reduction of the salt begins almost immediately, and the generation of nanoparticles is usually indicated by a change in the color of the reaction solution. The mechanism for the synthesis of NPs is shown in figure 5.

Plant-based synthesis is primarily divided into three stages: activation phase, growth phase, and termination phase [99, 100]. The activation phase is the first step in the process of recovering metal ions from their salt precursors using plant metabolites, which are biomolecules with reduction abilities. Nucleation of the condensed metal atoms occurs after the metal ions are changed from their monovalent or divalent oxidation states to zero-valent states [101]. The activation phase is followed by the growth phase (Ostwald ripening), in which the separated metal atoms coalesce to form metal NPs, with further biological metal ion reduction taking place. Nanoparticles accumulate laterally as they expand, forming a variety of morphologies such as cubes, triangles, rods, spherical, hexagons, pentagons, and wires [102]. The increased thermodynamic stability of NPs occurs as the growth stage progresses, while extensive nucleation can cause aggregation of synthesized

NPs, altering their morphologies. The termination phase (defining the final shape of NPs) is the final step in the green NP synthesis process. When the NPs are capped by plant metabolites, they get the most promising and stable morphology [94].

1.4. Clinical applications of gold and silver nanostructures/particles

Due to the extremely small size and multiple applications of NPs in the various fields, it has gained paramount importance in recent times [6]. NPs such as gold and silver have been recently used in bioimaging/ labeling, anticancer therapy, targeted drug delivery, agriculture, water treatment, nano fertilizers/ nano pesticides/ nano herbicides, catalytic activity, removal of pollutants, heavy metal ion sensing, acaricidal agent, anti-biofouling agent, anti-inflammatory, antileishmanial agent, antimicrobial agent, optics, mechanics, electronics and aeronautics, etc. [4, 6, 16]. The antimicrobial activity of NPs, can be divided into two categories: bactericidal and inhibitory. Bacterial cells die due to NPs activity in bactericidal action, whereas bacterial cell division is only prevented in inhibitory action without destroying any cells.

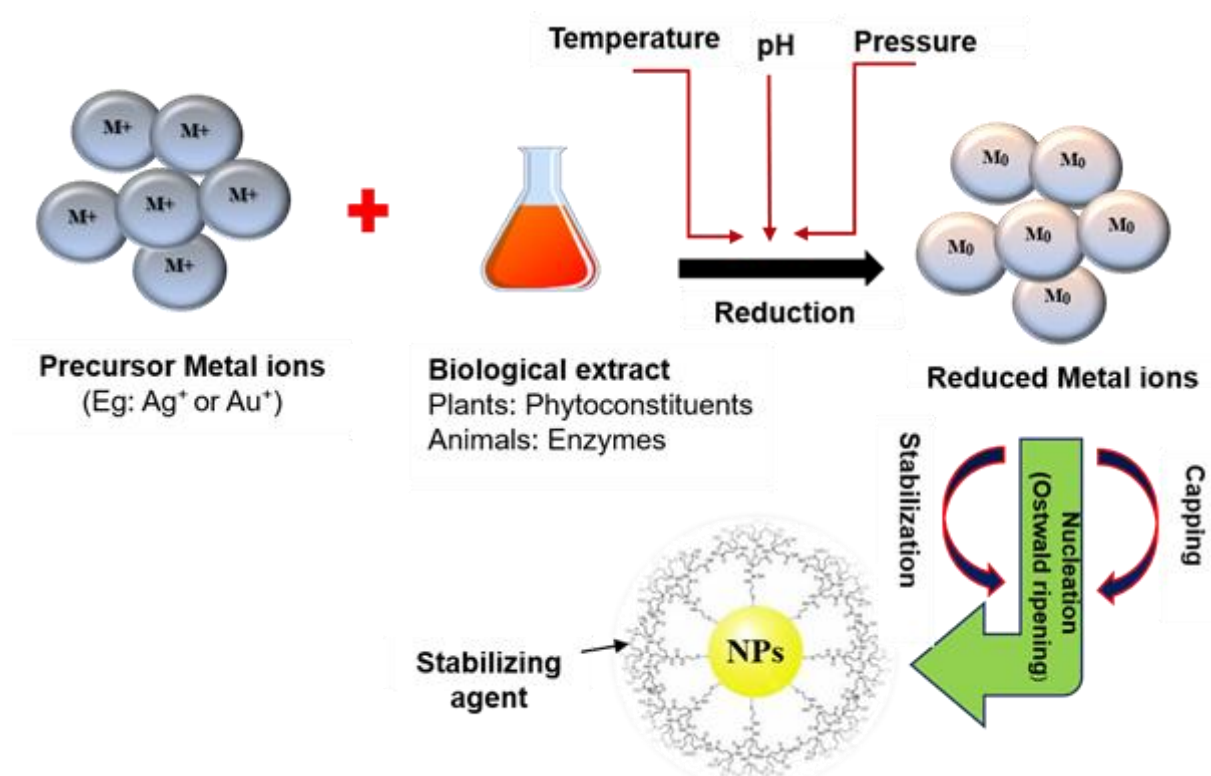


Fig. 5. Diagrammatic representation of mechanism of green synthesis of nanoparticles

Based on these two types of actions, the disc diffusion method, minimum inhibitory concentration determination, and minimum bactericidal concentration determination are three commonly used methods to study the antibacterial properties of any material.

1.4.1. Disc or Well Diffusion Method

The disc diffusion approach is widely used to demonstrate the antimicrobial activity of AgNPs. Nanoparticles are coated on the surface of suitable absorbent materials with uniform size and disc shape in this process. The disc is positioned over the targeted microbe's surface during the analysis, and the antimicrobial activity of AgNPs is shown by the development of an inhibition zone around the disc. Instead of a disc, small disc-shaped pits are formed on the surface of the agar plate for filling the test solution in the well diffusion process. The plates are inoculated with microbes and incubated under clean conditions in both processes to obtain a clear inhibition zone. The diameter of the inhibition zone around the well or disc can easily visualize the effect of AgNPs on the selected microbes [20]. The researchers used *Datura stramonium* leaf extract to make AgNPs and tested their antibacterial activity using the well diffusion method against Gram-negative (*E. coli*) and Gram-positive (*S. aureus*) bacteria. The antibacterial activity of AgNPs increases as the dose of stock solution increases, and better antibacterial activity of AgNPs was demonstrated against *E. coli* [103].

1.4.2. Minimum Inhibitory Concentration (MIC) or Minimum Bactericidal Concentration (MBC)

The minimum inhibitory concentration (MIC) is characterized as a concentration of nanoparticles that causes 100 % visible growth of the targeted microbe after 24 hours of incubation. The MIC is calculated by watching the bacteria develop in culture tubes containing the same amount of bacterial culture but different concentrations of AgNPs. MIC is defined as the minimum amount of AgNPs required to inhibit the growth of the bacteria [20].

MBC was determined using by using a fixed amount of AgNPs, which is normally higher than the MIC value. AgNPs are applied to a nutrient medium containing a growing bacterial inoculum, and the bacteria's growth is observed using UV-

Vis spectroscopy or a plate analyzer in terms of changes in the samples' optical density. Broth dilution measurements are also used for MIC and MBC analysis, in which a comparison of the result is made with the standard after the completion of the experiment [20].

1.4.3. Antimicrobial activity of gold Nanoparticles (AuNPs)

AuNPs are recently used to treat various diseases owing to their smaller size and greater therapeutic efficacy. In addition to biomedical applications of the AuNPs, it has also been used as antimicrobial agents in human, agriculture products, fertilizers, pathogens, and food. The antimicrobial activity of AuNPs is credited for its extremely small size, larger surface area, and its photothermic nature [5]. It is reported that the AuNPs interfere with the ionic interactions in the cell membrane, whereby the membrane gets damaged by pathogens. In addition, AuNPs often cause bacteria and fungi to die by increasing the expression of genes involved in the redox process [5]. However, the exact mechanism of antimicrobial activity of AuNPs is not still known. Some studies have also reported that it works by targeting sulfur or phosphorous base in the cell. When AuNPs are bound to a thiol group in an enzyme, the respiratory chain is disrupted, resulting in the production of a large number of free radicals and causing cell death. AuNPs, on the other hand, minimize ATP behaviors by reducing the interaction between tRNA and ribosomes [5]. It is also reported that AuNPs exhibit antimicrobial activity by hindering transmembrane hydrogen efflux. When the size of the NPs is much smaller than the size of the bacterial cell, however, a lower concentration of AuNPs will hinder bacterial growth by about 250-fold, causing the pathogen to die. NPs bind to the pathogen's cell wall and slow the cell process, causing death. By using the electrostatic force of attraction on the cell wall, AuNPs will attract DNA from bacteria. With the increased surface area and nano-size, the chances of cell death become higher. Table 5 shows the list of plants used for the synthesis of AuNPs and their antimicrobial activities [104]. AuNPs shows different action on gram-positive and gram-negative bacteria. Gram-positive bacteria have a thick peptidoglycan layer, whereas only a thin layer of peptidoglycan is present in gram-negative.

Table 5. Plants used for the synthesis of AuNPs and its antimicrobial activity [104]

| Sl. no | Plant name | Microorganisms tested | Method |
|--------|-------------------------------|----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|---------------------|
| 1 | <i>Acorus calamus</i> | <i>Escherichia coli</i> , <i>Staphylococcus aureus</i> | Agar well diffusion |
| 2 | <i>Trianthema decandra</i> L. | <i>S. aureus</i> , <i>Enterococcus faecalis</i> , <i>Streptococcus faecalis</i> , <i>E. coli</i> , <i>P. vulgaris</i> , <i>P. aeruginosa</i> , <i>B. subtilis</i> , <i>Yersinia enterocolitica</i> , <i>Klebsiella pneumoniae</i> and <i>C. albicans</i> | Well diffusion |
| 3 | <i>Solanum lycopersicum</i> | <i>S. aureus</i> and <i>P. aeruginosa</i> | Well diffusion |
| 4 | <i>Zizyphus mauritiana</i> | <i>S. aureus</i> | Luria medium |
| 5 | <i>Dracocephalum kotschyi</i> | <i>E. coli</i> , <i>Pseudomonas aeruginosa</i> and <i>Proteus vulgaris</i> | Agar cup-plate |
| 6 | <i>Mentha piperita</i> | <i>S. aureus</i> and <i>E. coli</i> | Müller-Hinton agar |
| 7 | <i>Solanum nigrum</i> | <i>Staphylococcus saprophyticus</i> , <i>Bacillus subtilis</i> and <i>E. coli</i> | Disc diffusion |
| 8 | <i>Nepenthes khasiana</i> | <i>E. coli</i> , <i>Bacillus</i> , <i>Candida albicans</i> and <i>Aspergillus niger</i> | Well diffusion |
| 9 | <i>Gloriosa superba</i> | <i>S. aureus</i> , <i>Streptococcus pneumoniae</i> , <i>Klebsiella pneumoniae</i> and <i>E. coli</i> | Disc diffusion |
| 10 | <i>Salicornia brachiata</i> | <i>Salmonella typhi</i> , <i>E. coli</i> , <i>P. aeruginosa</i> and <i>S. aureus</i> | Disc diffusion |

As a result, the peptidoglycan in Gram-positive bacteria is very strong and can penetrate AuNPs through the cell wall, whereas Gram-negative bacteria have a thin layer that can easily cause cell death. In addition to the thickness of peptidoglycan and size of AuNPs, other factors such as concentration, capping agents, and purification are accounted for its antimicrobial activity. It was also reported that the AuNPs synthesized from the plant extract are superior or show greater inhibitory potential compared to those synthesized using the chemical method. The antimicrobial activity of green synthesized AuNPs could be attributed to the AuNPs' synergistic effect with plant extract [105-107].

1.4.4. Silver Nanoparticles (AgNPs)

AgNPs inhibits bacterial growth by suppressing respiratory enzymes and electron transporting unit by blocking the functioning of DNA. The respiratory and permeability functions of the cell membrane are hindered when silver ions are bound to the cell surface of microorganisms. Silver interacts closely with the membrane's surface and penetrates deep into the bacterial cell, causing DNA to lose its replicability and eventually die [108].

Many researchers have reported that AgNPs have more antimicrobial activity towards gram-negative bacteria than gram-positive because of thin peptidoglycan layers and a beta-barrel protein called porins present on the cell wall

[109]. Furthermore, AgNPs with a high specific surface-to-volume ratio release Ag^+ , increasing their contact with microorganisms, improving silver ion dissolution, and improving the biocidal effect. Factors such as pH, temperature, AgNO_3 concentration, and microorganism type affect the antimicrobial activity of AgNPs [20].

The study was carried out by Ruiz-Baltazar et al. to determine the antimicrobial properties of AgNPs synthesized via *Melissa officinalis* leaf extract against *S. A* and *E. coli*. The disc diffusion method and the MBC/MIC test were used to investigate the antibacterial activity. Both AgNPs solutions of various concentrations (ranging from 1 mM to 20 mM) were found to have an inhibitory effect. In both cases, the antibacterial properties of prepared AgNPs synthesized using a green synthesis method were verified by the inhibition zone results. In addition, it was also reported that with an increase in the AgNPs concentration, the antibacterial properties got enhanced, and better result was exhibited against *E. coli* than *S. aureus*. However, antibacterial properties were exhibited by AgNPs against both the microorganisms [110]. Photoinduced green synthesis and antimicrobial evaluation of polycaprolactone/ curcumin/ grape leaf extract-Ag hybrid nanoparticles (PCL/ Cur/ GLE-AgNPs) were reported by EL-Sherbiny et al. In comparison to GLE, GLE/ AgNPs, and PCL/ Cur/ GLE plain NPs, bacterial strains incubated with high concentrations of PCL/ Cur/ GLE-Ag hybrid NPs showed a high bactericidal

impact. PCL/ Cur/ GLE-Ag hybrid NPs have a high surface area, enabling them to make direct contact with bacterial cells and free radicals formed by AgNPs, which can damage bacterial cell membranes [111].

The green synthesis of AgNPs from silver acetate was reported by Pethakamsetty et al., who used methanolic root extracts of *Diospyros sylvatica* (a member of the Ebenaceae family) as a reducing agent. The synthesized AgNPs were reported to exhibit mild activity against *Proteus vulgaris* and *Streptococcus pyogenes*, but moderate activity against *E. coli*, *Staphylococcus aureus*, and *Klebsiella pneumoniae*. However, *Bacillus subtilis*, *Bacillus pumilis*, and *Pseudomonas aeruginosa* were shown the most activity by AgNPs [112]. The green synthesis of AgNPs by tissue extract from weaver ant larvae has been documented by Khamhaengpol and Siri (*Oecophylla smaragdina*). The synthesized AgNPs exhibited potential antibacterial activity against both *E. coli* and *S. aureus*. In contrast to other published literature, the MBC and MIC against both bacteria were found to be the lowest [113].

1.5. Factors affecting green synthesis

1.5.1. Particular Method or Technique

There are various methods to synthesize NPs. The methods or techniques include physical method (Mechanical procedure), chemical method (using chemicals), and biological method, which include the use of organic or inorganic chemicals, plants, and microorganisms for the synthesis of NPs. Each method is associated with certain benefits and limitations. Physical and chemical methods are associated with extensive use of energy and

toxic chemicals, which are harmful to both people and the environment. Biological methods for nanoparticle synthesis, on the other hand, use nontoxic and environmentally safe materials in combination with green technology, making them more environmentally friendly and more suitable than conventional methods [121, 122].

1.5.2. Effect of concentration on synthesis

An increase in concentration increases the rate of collision of reactant particles and hence facilitates the rate of reduction of metal ions into metal nanoparticles. Although less concentration (3.5×10^{-6} M) of flavonoid derivatives has the ability to reduce metal ions, they do not have the ability to act as capping agents and stabilizers as compared to the high concentration (3.5×10^{-2} M) of phytochemical constituents (i.e., flavonoids) or enzymes. A higher concentration of flavonoid produces AuNPs with small size. It also indicates that when a high concentration of flavonoid derivatives was used, it stimulated strong interaction between the protective flavonoid derivatives and the surface of AuNPs, preventing AuNPs from sintering, thus resulting in size reduction of AuNPs [123].

1.5.3. Choice of best organism

In order to have an efficient and effective synthesis of NPs, researchers have given core importance to intrinsic properties of organisms such as enzymatic activities and biochemical pathways. It was also reported that plants with great potential for heavy metal accumulation and detoxification are the best candidates for the synthesis of metal nanoparticles [4].

Table 6. Plant extract-mediated green synthesis of AgNPs with antimicrobial activity.

| Sl. no | Plant Extract | Targeted pathogens | Ref. |
|--------|-------------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------------------|------------------------------|
| 1 | <i>Citrus limetta</i> peel | <i>Micrococcus luteus</i> , <i>Streptococcus mutans</i> , <i>Staphylococcus epidermidis</i> , <i>S. aureus</i> , <i>E. coli</i> , <i>Candida spp.</i> | (Dutta et al., 2020) |
| 2 | <i>Luffa acutangula</i> leaf | <i>E. coli</i> , <i>Saccharomyces cerevisiae</i> | (Kaushal et al., 2016) |
| 3 | <i>Parkia speciosa</i> leaf | <i>E. coli</i> , <i>S. aureus</i> , <i>Pseudomonas aeruginosa</i> , <i>Bacillus subtilis</i> | (Ravichandran et al., 2019) |
| 4 | <i>A. indica</i> leaf | <i>S. aureus</i> , <i>E. coli</i> . | (Ahmed et al., 2016) |
| 5 | <i>Gomphrena globosa</i> leaf | <i>S. aureus</i> , <i>B. subtilis</i> , <i>M. luteus</i> , <i>E. coli</i> , <i>P. aeruginosa</i> , <i>Klebsiella pneumoniae</i> . | (Tamilarsi & Meena, 2020) |
| 6 | <i>Pedaliium murex</i> leaf | <i>E. coli</i> , <i>K. pneumonia</i> , <i>Micrococcus flavus</i> , <i>P. aeruginosa</i> , <i>B. subtilis</i> , <i>Bacillus pumilus</i> , <i>S. aureus</i> | (Anandalakshmi et al., 2016) |
| 7 | <i>Musa acuminate</i> peel | <i>B. subtilis</i> , <i>S. aureus</i> , <i>P. aeruginosa</i> , <i>E. coli</i> . | (Ibrahim, 2015) |

1.5.4. pH

pH is the major factor that has a significant influence on the synthesis of NPs using the green synthesis techniques. Many researchers have discovered and reported that the pH of the solution medium has effects on the texture and size of the synthesized nanoparticles [124, 125]. As a result, changing the pH of the solution media will regulate nanoparticle size. A study demonstrated the effect of pH on the shape and size of the synthesized silver nanoparticle [126].

1.5.5. Temperature

Temperature is another significant factor that influences nanoparticle synthesis in all three methods, namely physical, chemical and biological methods (green synthesis). The physical method necessitates the maximum temperature ($>350^{\circ}\text{C}$), while chemical methods necessitate temperatures below 350°C . In certain cases, green technology nanoparticle synthesis requires temperatures of less than 100°C or ambient temperature. The nature of the nanoparticles produced is determined by the temperature of the reaction medium [127].

1.5.6. Pressure

The pressure is an essential factor for the synthesis of NPs. The shape and size of the synthesized nanoparticles are influenced by the pressure applied to the reaction medium [128]. At ambient pressure, the rate of metal ion reduction using biological agents was reported to be much faster [129].

1.5.7. Particle Shape and Size

The size of nanoparticles plays an important role in determining their properties. The melting point of nanoparticles, for example, has been shown to decrease as their size approaches the nanometre scale [130]. Nanoparticles of various configurations have similar energies, making shape transformation easy. The type of energy typically used in nanoparticle analysis causes the nanoparticle to change shape [131]. The chemical properties of synthesized nanoparticles are greatly influenced by their complex nature and form. The type of energy typically used in nanoparticle analysis causes the nanoparticle to change shape. The chemical properties of synthesized nanoparticles are greatly influenced by their dynamic nature and form [132].

1.5.8. Time

The length of time the reaction medium is incubated has a significant impact on the quality and form of nanoparticles synthesized using green technology [133]. Similarly, the properties of synthesized nanoparticles changed over time and were heavily affected by factors such as the synthesis process, exposure to light, storage conditions, etc. [134, 135]. The study also reported that the green synthesis via plant extract is much faster compared to microorganisms (bacteria, fungi, yeast) as the latter requires a longer incubation period for the synthesis. Variations in time may occur in a variety of ways, including particle accumulation as a result of long-term storage; particles can shrink or expand as a result of long-term storage; they may have a shelf life, and so on, all of which affect their potential [136].

1.5.9. Preparation Cost

The costs associated with nanoparticle synthesis must be regulated and monitored in order to promote the future application of nanoparticles in modern-day applications. As a result, the cost-effectiveness of the manufacturing process has a significant impact on nanoparticle synthesis. Despite the fact that the chemical method of synthesis produces a high yield in a limited amount of time, it is not cost-effective. As a result, chemical and physical synthesis of nanoparticles may be limited, whereas biological synthesis of nanoparticles is less expensive and can be done on a large scale [15].

1.5.10. Pore Size

The porosity of the synthesized nanoparticles significantly affects the quality and applications of the nanoparticles. The pore size of nanoparticles affects the drug loading and drug release characteristics. The larger the pore size, the drug loading capacity will be higher, release rate will be faster and so the therapeutic efficacy will be increased [137].

1.5.11. Environment

The nature of the synthesized nanoparticles is heavily influenced by the surrounding environment. A single nanoparticle can quickly become core-shell nanoparticles in many environments by absorbing materials or reacting with other materials in the environment through

oxidation or corrosion [138]. The synthesized nanoparticles form a coating in a biological environment, making them thicker and larger in size [139]. The physical structure and chemistry of the synthesized nanoparticles are also affected by the environment. There are a few examples of how the environment affects the composition of nanoparticles that have been synthesized. As the atmosphere of the zinc sulfide nanoparticles was modified from wet to dry, the crystalline nature of the nanoparticles changed instantly. Another example also supports that the presence of peroxide in the solution modifies the chemical nature of cerium nitrate, which in turn indicates that the environment has an effect on the nature of NPs [134].

1.5.12. Proximity

In most cases, when an individual or isolated nanoparticles come into contact with or near the surface of other nanoparticles, their properties are altered. This shifting behavior of nanoparticles can be used to create more precise nanoparticles. The proximity effect of nanoparticles has numerous effects, including substrate interactions, particle charging, and nanoparticle magnetic properties [140].

1.5.13. Other Factors

The plants and microorganisms are rich in secondary metabolites, which have the ability to act as both reducing and stabilizing agents for the synthesis of NPs. But the composition of the various metabolites varies with the type of plant, part of the plant, and the procedure used for the extraction process [141]. Not unlike plants, various microorganisms produce different intracellular and extracellular enzymes that influence nanoparticle synthesis in different amounts [142]. Furthermore, the process used to purify the synthesized nanoparticles may have an effect on the quantity and consistency of the nanoparticles.

1.6. Recent advances

1.6.1. Method or substrate used for green synthesis

1.6.1.1 Biopolymers

Polymers synthesized by living organisms are called biopolymers. They can also be synthesized from renewable resources by polymerization. They are biodegradable polymers made up

primarily of organic compounds. Proteins, carbohydrates, DNA, RNA, lipids, peptides, and polysaccharides are examples of biopolymers. Functional groups such as carboxylic, amide I and II, ketone, aldehyde present in the biopolymers assist in the production of NPs. [143-145]

Chitosan is one of the most commonly used biopolymers as a reducing agent in green chemistry. Deacetylation of chitin, a naturally occurring polymer, produces chitosan, a cellulose-like linear polysaccharide. It can be present in a wide variety of insects, fungi, yeasts, and marine invertebrates. [146]

The study was carried out by Esther et al. to investigate the impact of different preparation methods on the size and shape of NPs. In that study, two different preparation methods were employed. The first method involved dissolving chitosan in a boiling HAuCl_4 solution, while the second method involved dissolving a HAuCl_4 solution in hot boiling chitosan. The nucleation rate was rapid and standardized when the HAuCl_4 solution was applied directly to the boiling chitosan solution. However, the rate of nucleation was not uniform when chitosan solution was poured in hot HAuCl_4 as the metal ions were absorbed on the matrix and slowly reduced to the nuclei. NPs of various shapes and sizes were made [147]. In the presence of a plant-based extract, chitosan not only acted as a reducing agent, but also acted as a stabilizing agent.

Furthermore, Saha et al. conducted a study with chitosan and black pepper extract. When both chitosan and plant extract were used at the same time, the stability was improved. They suggested that chitosan acted primarily as a stabilizer by interacting electrostatically with the negatively charged AuNPs and positively charged polymeric network. [148]

1.6.1.2 Enzymes

Due to the numerous disadvantages of microbial bio-reduction processes, researchers have proposed that enzymes could be one of the best possible routes for the synthesis of NPs. They conclude that the existence of enzymes, proteins, carbohydrates, and bio membranes in microbes allows them to minimize AuNPs. However, studies have shown that it takes a longer incubation period for microbes and thus has a slow rate of synthesis compared to the plant-based synthesis [149]. Enzymes are also available

in pure form on the market. Purification and processing of NPs would be made easier as a result of this. Nisin peptides, a class-Ia bacteriocin, can be used to synthesize AuNPs via an enzymatic path. *Lactococcus lactis* produces a peptide that contains 34 amino acids. The NPs were spherical in shape and measured 25 nm in diameter. XRD review validated this finding. [150].

1.6.1.3 Green synthesis from vitamins

Vitamin B2 (as reducing and capping agents) was used to create a green mixture of Ag and palladium nano-spheres, nanowires, and nanorods. For the synthesis of nanowires and nanorods, vitamin B2 is used as a reducing agent. This is a novel approach in the field of green nanotechnology in that it proposes the use of natural agents in the development of the field, such as their effects on various tumor cells [151]. Ascorbic acid is used as a capping and reducing agent, and chitosan is used as a stabilizing agent. Since chitosan bonds with metal ions, the concentration of NPs is directly proportional to the chitosan concentration used [152].

The use of ascorbic acid as both a reducing and capping agent for the synthesis of NPs with uniform size was reported. Antioxidants that are water-soluble, such as ascorbic acid, seem to be responsible for the reduction of Ag NPs in *Desmodium triflorum*. Plants generate a significant amount of H^+ ions along with Nicotinamide adenine dinucleotide (NAD) during glycolysis, which acts as a powerful reducing agent and appears to be helpful in the formation of AgNPs. [153]

1.6.1.4 Microwave-assisted synthesis

By altering the parameters like surfactants, solvent, and metallic precursor, nanowires, tubes, and dendrites can be produced. Spherical NPs can also be produced by this synthetic method [154]. For the production of Ag NPs, carboxyl methyl cellulose sodium is used as a reducing and capping agent. MW union supports homogeneous heating and quick nucleation of noble metal NPs, in contrast to general heating treatment [155]. By using red grape pomace as a reducing agent, a quick NPs generation method (within seconds) for the synthesis of Ag, Au, palladium, and Pt in an aqueous medium by MW irradiation at 50 W was recorded [156].

1.6.2. Recent applications of Au and Ag

1.6.2.1 Water treatment

reduce the quality of pure drinking water and crop irrigation and, in turn, pose various health hazards to humans and other living organisms [157]. Due to the high surface area to volume ratio, enhanced catalytic activity, and chemical stability of green synthesized AgNPs and AuNPs, they serve as the best candidates for water treatment, purification, water monitoring, and agriculture wastewater treatment [158, 159]. Moreover, researchers have learned that NPs have great potential to detect toxic substances and heavy metals such as lead, cadmium, and mercury by incorporating sensors for quick detections [160, 161]. Pesticides present in the water are absorbed onto the NPs, retained on the surface, and then precipitate in the form of the complex. As a result, the water gets purified [162, 163]. The leaf extract of *Mussaenda glabrata* was used to synthesize AgNPs and AuNPs by Francis et al. to evaluate the antimicrobial activity against the pathogenic microorganisms. The NPs exhibited significant antimicrobial activity against the *E. coli*, *P. aeruginosa*, *Penicillium chrysogenum*, and *A.niger*. [164]

1.6.2.2 Targeted drug delivery

Due to the high surface area and greater biocompatibility of AuNPs, it has gained major attention in targeted drug delivery systems. NPs are bonded to the drug by either a covalent or non-covalent bond for its delivery at the targeted site. The drug release occurs by photothermal, triggered or ultrasonic method [165]. The drug release is dependent on parameters like pH, temperature, and biomolecule levels in the body. Targeted delivery using AuNPs has gained major attention in the treatment of cancer or tumor tissues than normal cells. Tumor cells, on average, are more temperature-sensitive than normal cells. As a result, drugs capped on AuNPs may be desorbed on the targeted spot during the heating phase, causing the drug and AuNPs to lose their bond. The drug release of glutathione-AuNPs in tumor cells is based on this theory [166]. Another method requires the deposition of layers of AuNPs on drugs using ultrasound radiation. The bond between AuNPs and drugs is broken when these layered AuNP drugs are exposed to ultrasound radiation. Finally, the drug will be released onto AuNPs' surface. One study reported

that antibiotics coated with AuNPs demonstrated bacterial activity against multidrug-resistant bacteria [167]. AuNPs demonstrated potent antibacterial activity against Gram-positive (*Staphylococcus aureus*) and Gram-negative (*Escherichia coli*) bacteria using ampicillin-AuNPs as a result of this application. All of the research that used drug-coated AuNPs found that they greatly enhanced drug delivery while also improving performance for a variety of diseases. [167, 168]

1.6.2.3 Cancer treatment

Green synthesized NPs of different designs, compositions, shapes, and sizes have potential against many cancer cells lines through antimetastatic [169], antiproliferative [170] cytotoxicity [171], and apoptosis [172]. Once the tumor is attached to the main blood circulation system, NPs then exploit several features of the newly developed vasculature to effectively attack it. Tumor cells are supplied by blood capillaries that perfuse the tissue's cells, where NPs can passively accumulate or anchor to biomarkers overexpressed in tumor cells by targeting moieties. NPs may serve as therapeutic agents, enhancing radiotherapy, silencing genes, inducing hyperthermia, and delivering drugs to cause tumor cell death, as well as imaging enhancers or contrast agents to aid in real-time tracking of therapeutic effects.

Patra et al. used *Butea monosperma* leaf extract to synthesize Ag and AuNPs, in that leaf extract served as a stabilizing, capping, and reducing agent. The AgNPs and AuNPs generated are biocompatible with cancer cell lines and normal endothelial cells, as well as stable in the biological buffer. The synthesized NPs, in combination with doxorubicin, inhibited cancer cell growth better than the pristine compound [173].

1.6.2.4 Nano-fertilizers

To synchronize nutrient release with plant uptake, AuNPs and AgNPs-based nano fertilizers have been developed. This method helps to preserve soil fertility by reducing nutrient loss, soil and groundwater pollution, and chemical reactions between water, soil, and microorganisms that turn them into toxic substances for plants [174-176]. The study was carried out by Kang et al. to demonstrate the effect of AgNPs fertilizers on the

red ginseng shoot. In that study, they applied 5 mg/L AgNPs fertilizer suspension to ginseng shoot 3 times per day at 14 days intervals. They discovered that the nano fertilizer had increased the ginsenoside content after harvesting [177].

Nano encapsulated herbicides and pesticides exhibited improved characteristics in terms of stability, solubility, specificity and permeability as the nano structure coating protects the active substance from early degradation and provides longer periods of pest control [176]. Furthermore, phytopathogens that cause plant disease, such as bacteria and fungi, can be controlled by spraying an NPs solution directly on crops, seed, or foliage to prevent pathogen invasion [178]. El-Aziz et al. synthesized AgNPs from *Fusarium solani* in order to test their effect on grain-borne fungi. According to the findings, 4 percent sprayed NPs solutions resulted in a 0% frequency of fungal pathogens [179].

1.6.2.5 Metal sensing

Heavy metals such as Pb, Fe, Cd, Ni, Hg, Mn, Cr, Co, and Zn are the major pollutants in soil, air, and water. Some heavy metals such as lead, cadmium, copper, and mercury ions have the enormous potential to cause toxicity at trace ppm levels [180, 181]. As a result, identifying toxic metals in the biological and marine ecosystem has become critical for effective remediation [182, 183]. In multi-element analysis, traditional techniques focused on instrumental systems usually have excellent sensitivity. However, the experimental setups are costly, tedious, skill-required, and non-portable. Metallic nanoparticles have been preferred for the detection of heavy metal ions in contaminated water systems due to their tunable size and distance-dependent optical properties [184, 185]. The mercury and lead ions (Hg^{2+} and Pb^{2+}) were selectively sensed by AgNPs synthesized from fresh and dried mango leaves. In another study, AgNPs were synthesized from pepper seed extract and green tea extract to demonstrate selective sensing for Hg^{2+} , Pb^{2+} , and Zn^{2+} ions [186].

1.6.2.6 HIV treatment

The immune system of our body is capable of removing some viral infections itself. However, some of the infections such as herpes, hepatitis, HIV, are long-lasting, persistent, and cannot be

removed. AuNPs are used as a delivery scheme for the improved effectiveness of the anti-HIV activity. The free AuNPs are unproductive against HIV infection. Kesarkar et al. (2012) [187] found that stabilized AuNPs in amino acid L-cysteine resulted in the enhanced azidothymidine delivery against the HIV-1Ba-L virus *in-vitro*.

AuNPs are being used as a delivery system to increase the efficacy of anti-HIV activity. AuNPs exhibit anti-HIV activities due to poly-anionic shallow, which allows binding with the positively charged amino acid at the binding position of glycoprotein GP120 (or gp120). The mechanism of inhibition takes place, and the reverse transcriptase enzyme of HIV-1 protein is blocked by AuNPs. A study carried out by Garrido et al. reported that conjugated AuNPs have better anti-HIV activities than free AuNPs, which don't exhibit anti-HIV activities.

1.7. Conclusion and future prospective

Green synthesis of nanoparticles has gained enormous importance in the field of nanoscience over the conventional methods of nanoparticles synthesis [1]. Green synthesis is an eco-friendly, cost-effective, energy-efficient, rapid and effective method. Generally, the synthesis of a nanoparticle is broadly categorized into two major divisions; top-down approach and bottom-up approach. Green synthesis is categorized under the bottom-up approach [6]. Green synthesis involves the use of plants and microorganisms for the synthesis of nanoparticles. However, in the green synthesis, synthesis via plant is preferred over microorganisms as the phytochemical present in the plant is an efficient and rapid process as extracellular and intracellular enzymes present in the microorganisms require a longer incubation period for the synthesis [133]. The studies have reported that intracellular and extracellular enzymes present in microorganisms have the ability to reduce precursor metal salt ions into their respective nanoparticles [87].

In terms of plant, the synthesis process is reported to proceed via three steps, namely activation phase, growth phase, and termination phase [99, 100]. Phytochemicals present in the plants also have the ability to reduce metal ions into nanoparticles. Owing to its high therapeutic efficacy and application in various fields, gold and silver nanoparticles have gained popularity over other nanoparticles. They are used in

bioimaging/labeling, anticancer therapy, targeted drug delivery, agriculture, water treatment, nano fertilizers/ nano pesticides/ nano herbicides, catalytic activity, removal of pollutants, heavy metal ion sensing, acaricidal agent, anti-biofouling agent, anti-inflammatory, antileishmanial agent, antimicrobial agent, optics, mechanics, electronics and aeronautics [4, 6, 16]. The antimicrobial activity of gold and silver nanoparticles is determined by disc or well diffusion method, minimum inhibitory concentration, and minimum bactericidal concentration [20]. Various invitro studies have reported the antimicrobial activity of AgNPs and AuNPs against various microbes. Studies also found out that the antimicrobial activity was concentration-dependent [123]. In addition, studies have also reported that the AgNPs have less antimicrobial activity against gram-positive bacteria than gram-negative bacteria [109].

The green synthesis of gold and silver nanoparticles is affected by the parameters such as techniques or methods used, the concentration of substrate, pH, temperature, pressure, time, pore size, preparation cost, particle shape and size, pore size, types of reducing agent and capping agent used, the proximity of other nanoparticles, environment and other factors [15].

Recently, the green synthesis of NPs has gained major attention and importance in the field of nanoscience owing to its diverse applications and relatively small size. However, exact mechanisms for the synthesis are not clearly known. Further studies and research on AuNPs and AgNPs are needed to clearly understand the exact mechanisms of green synthesis. In addition, studies must be focused on the determination of main reducing components so that more precise and suitable methods can be established. Detailed synthesis mechanism should be mastered before implementing it into different fields. NPs hold a brighter future in the fields of medicine, technology, industry, and chemistry. Moreover, NPs hold the greatest potential in biomedical applications. Due to the multi-surface functionality of NPs, it will serve as the best candidate for the targeted delivery of drugs for the effective treatment of diseases with minimal side effects. Also, AuNPs are capable of emitting fluorescence, which can, in turn, be used as the biomarker for diseases. As a result, the disease can be effectively diagnosed. Most of the in-vitro

studies concluded that they are non-toxic and effective. However, more in vivo studies are needed to be conducted to clearly understand the toxicity of NPs, duration of NPs in the body, time of excretion, and effects of residual NPs in the organism. Also, the dosage of NPs needs to be studied so that right and suitable quantities of NPs can be delivered to the patient. Therefore, more detailed and extensive studies and research are needed to be carried out to completely understand and harness the latent potential of AuNPs and AgNPs in the near future.

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